

Environmental Importance of Mg

Mg is ubiquitous in soils and water.
Indigenous and anthropogenic sources.
Mg-P occurs in manures; is more soluble than most Ca-P.
Mg-P in dairy manure can be preempted by high dietary Ca.
Mg can inhibit stable Ca-P crystallization in soils, however ...
Mg can also ENABLE Ca-P crystallization.

Objectives

We present data from several studies that –

- Document Mg-P in dairy manure and its effect on water extractable P.
- Specify conditions for inhibiting and enabling effects of Mg on Ca-P.

We also propose a mechanism for Ca-P “enabling” effect in Ca–Mg–CO₃–PO₄ system.

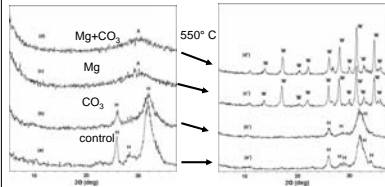
Materials & Methods

Materials: Soils, manures, reactor seed grains (quartz), fluidized-bed reactor.

Methods: Fluidized-bed reactor for P recovery, chemical analyses, x-ray diffraction (XRD), scanning electron microscopy (SEM), energy-dispersive x-ray fluorescence elemental spectroscopy (EDS).

Mg in Dairy Manure: Inhibitor or Preemptor of Ca-P?

Mg as Potential Inhibitor of Ca-P Crystallization



XRD patterns of precipitated solids from solutions containing P and the labeled components.

Minerals with peak labeled: A, amorphous Ca-Mg-P; H, hydroxyapatite; W, whitlockite.

Mg inhibits formation of the most stable crystalline P form (H).

Mg results in formation of amorphous P phase.

Mg incorporation produces whitlockite stoichiometry.

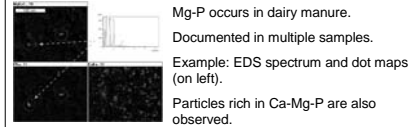
Whitlockite crystallizes with heat.

Solutions where amorphous Ca-Mg-P forms have higher P.

This behavior also shown for simulated dairy soil solutions.

Whitlockite is evident in ashed dairy manure.

Mg as Potential Preemptor of Ca-P Crystallization



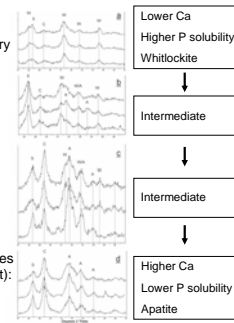
Recent work showed that increasing dietary Ca reduced dairy manure P solubility (dietary P constant).

Mg-P formed in cows fed diets with lower (but nutritionally adequate) Ca availability.

Ca-P formed with higher (but safe) Ca availability.

In effect, Mg-P preemptively formed at lower Ca availability, rendering P in manure more soluble.

See XRD of ashed dairy samples with increasing dietary Ca (right): A, apatite; W, whitlockite; C, calcite.

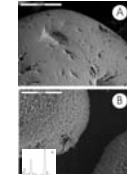


Mg as Enabler of Ca-P

P Recovery “Discovery”

Ca-P recovery from flushed dairy manure in fluidized-bed reactor foiled by CaCO₃ precipitation.

Recovery enabled by MgSO₄.



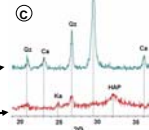
(A) SEM image of clean quartz grain used for seed material in a P-recovery reactor, showing smooth surface;

(B) SEM image of quartz seed grain after P recovery via pH elevation + MgSO₄, showing grain surface covered with precipitate dominated by Ca & P (insert).

(C) XRD of seed grain coatings

Without Mg, only CaCO₃ (Ca) precipitated

With Mg, poorly-xline apatite (HAP) precipitated



Why would Mg enable Ca-P precipitation?

Clues: High alkalinity and pH elevation via alkali.

Proposed Enabling Mechanism

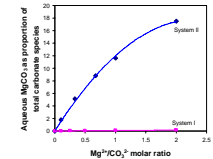
Ca–Mg–CO₃–PO₄ Systems

At circum-neutral pH (System I below), (CO₃) is relatively low; (MgCO₃)_{aq} is also low even at high (Mg).

At elevated pH in P recovery (simulated by System II) (MgCO₃)_{aq} is relatively high proportion of total carbonate species at higher (Mg).

Formation of (MgCO₃)_{aq} prevents:

- CO₃ from precipitating as CaCO₃.
- Mg from inhibiting apatite nucleation.



Caveat: Higher NH₄ could produce struvite rather than Ca-P or Ca-Mg-P.

References

- Josan, M.S., V.D. Nair, W.G. Harris, and D.A. Herrera. 2005. Associated release of Mg and P from active and abandoned dairy soils. JEQ 34:184-191.
- Cao, X., W.G. Harris, M. Josan, and V.D. Nair. 2007. Inhibition of Ca-P precipitation under environmentally relevant conditions. Sci. Total Environ. 383:205-215.
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Acknowledgments

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