

# Oxidation in Soil of CaSO<sub>3</sub> from Flue Gas Desulfurization Product and Effect on Corn Establishment

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### ABSTRACT

Application of gypsum (CaSO<sub>4</sub>2H<sub>2</sub>O) to agricultural soils has a long history. Hannebachite (CaSO<sub>3</sub>+H<sub>2</sub>O) is a product created when coal is burned and sulfur is scrubbed from the flue gases. It is considered inferior to gypsum as an agricultural amendment because it can negatively impact plant growth, especially in acid soils, and has poor handling properties. For agricultural use of FGD-CaSO<sub>3</sub>, it must first be oxidized to sulfate (SO<sub>4</sub><sup>2</sup>) in soils before crops are planted. However, there is little information about the oxidation rate of sulfite (SO<sub>3</sub><sup>3</sup>) under field conditions. An FGD-CaSO<sub>3</sub> was applied at rates of 0, 1.12 and 3.36 Mg ha<sup>-1</sup> to the surface of an agricultural soly (Wooster silt loam, Oxyaquic Fragiudalf). Oxidation of SO<sub>3</sub><sup>2</sup><sup>2</sup> was determined by measuring SO<sub>4</sub><sup>2</sup> in the surface soil (0-10 cm) on 0, 1, 2, 3, 8, and 14 days after application. Corn was planted on the same days as samples were analyzed for SO<sub>4</sub><sup>2</sup> and corn emergence and growth were measured. It is safe for plants if FGD-CaSO<sub>3</sub> is applied to field surface two weeks before planting.

#### INTRODUCTION

Plants take up S mostly in the form of SQ.<sup>2</sup>. At low soil pH, SQ.<sup>2</sup> may be converted to sulfur dioxide (SQ.) which is toxic to plants (Rttchey et al., 1999). However, CaSQ, can rapidly oxidize to CaSQ, in oxygenated environments (Pasituk-Bronkowska et al., 1992). The reaction rate is strongly affected by concentrations of dissolved SQ.<sup>2</sup> and Q<sub>2</sub>, pH, and temperature (Lancia and Musmarra, 1999). Hao and Dick, 2000; Lee et al., 2007). For agricultural use, FGD products are often applied to the surface of the fields. However, there is no report on the oxidation of FGD-CaSQ, and its effect on the emergence and growth of plants under field conditions. The objectives of this study were to determine the oxidation rate of FGD-CaSQ, at the soil surface of an agricultural lield and to evaluate the impact of FFGD-CaSQ, on core mergence and growth.

#### MATERIALS AND METHODS

A field study was conducted on an agricultural soil (Wooster silt barn, Oxyaquic Fragiludaf) focation fare Wooster, O.H. Selected characteristics of the soit ap represented in Table 1, FSD-CaSO, was obtained from American Electric Power Company, Conswille, OH, and FSD-Oxynsum was obtained from Cinergy Corporation, Cincinnati, OH (Table 2), FSD-CSSO, was applied at rates of 0, 112, and 33 M (bh at to the soil surface, and FSD-poysum was applied at only a single rate of 1.12 Mg hat a a positive comparison. Rates used were normal field recommended rates required to improve soil physical and chemical properties. These treatiments were applied to 1 ×1 m plots arranged in a randomized block with four replicates.

Table 1. pH and concentrations of selected elements extracted from										Table 2. Characteristics of the FGD-CaSO <sub>3</sub> and FGD-gypsum used.					
expe	rime	ntal fie	ld soil	(0-20 c	m) by	/ Mehli	ch-III.						Parameter	FGD-CaSO <sub>3</sub>	FGD-gypsum
<u>рН</u> 7.1	  50	<u>к</u> 111	<u>Ca</u>	Mg 311	<u>S</u> 51	B m 0.43	<u>Cu</u> g kg <sup>-1</sup> 1.5		<u>Mn</u> 94	<u>Mo</u> 0.03		AI 775	Sulfur component (%) SO <sub>3</sub> -S SO <sub>4</sub> -S	17.0 0.7	0 18.5
2007 (0-10 ratio) chroi extra (ICP) each The sowin after	. On with mato cts of emi plot rates ng, a sowi	days were doub graph on day ission on th of co ind dr ing. C	iment 1 0, 1, 2 collected ble deid y. Con 14 we spectre same y weigh orn pla orn pla	e, 3, 8 ted an onized acentra ere an rometre e days ergend ht of c ants se	and f nd SC water ations alyze y. Th s as s ce we orn p own c	14 after D <sub>4</sub> <sup>2</sup> was a of variation of va	er trea as ex anal rious nduct seed es we asure was o 14 v	atmen tracte yzed l elem tively ds of c re ana ed on detern were d	ts, so d (2:3 by ion ents i couple orn w alyzed day 8 hined ried i	il sam soil to n the ed pla vere so d for S after on da n a 65	ples b wat sma bwn ii $O_4^{2^*}$ . y 21		Element (mg kg ') P K Ca Mg S B Cu Cu Fe Mn Mn Mo Zn Al	6.3 278 247000 7480 177000 261 - <0.20 725 23.9 0.40 17.9 1660	40.80 75 222000 105 185000 4.7 <0.20 122 0.05 0.32 18.5 251

elements were determined by ICP emission spectrometry after

HCIO, :HNO, digestion.

## RESULTS AND DISCUSSION

Elemen

Concentrations of water-soluble SO<sub>4</sub>-S in the soil surface layer (0-10 cm) were increased by FGD-CaSO<sub>3</sub> or FGD-gypsum treatment (Fig. 1). Concentrations of SO<sub>4</sub>-S in soil treated with 1.12 Mg ha<sup>-1</sup> FGD-CaSO<sub>3</sub> were similar to those treated with FGD-gypsum at II days measured. This indicates the SO<sub>3</sub><sup>-2</sup> in the FGD-CaSO<sub>3</sub> was rapidly oxidized after surface application.

Ca and S concentrations in the soil. were significantly increased by FGD-CaSO, or FGD-gypsum treatment, as were also water-soluble K and Mg (Table 3), The Ca jons have greater affinity for exchange sites on soil particles than K and Mg ions and the Ca in the FGD-CaSO, and FGD-gypsum displaced and mobilized K and Mg. As expected, there were higher concentrations of Mg, B, and Mn in soil treated with FGD-CaSO<sub>2</sub> than with FGD-avpsum due to their higher concentrations in FGD-CaSO<sub>2</sub> (Table 2). Application of FGD-CaSO, or FGDgypsum significantly decreased the concentrations of water-soluble P. Fe. Mo. Zn and Al. These decreases were attributed to leaching losses caused by Ca replacement and precipitation reactions. Aluminum is a major factor limiting crop yield in many areas of the world. Surface application of gypsum has become a recommended procedure for Al amelioration. Our results indicate that FGD-CaSO<sub>3</sub>, like gypsum, is a good product for Al amelioration.



Fig. 1. Changes in water-soluble SO₄-S in the surface 10 cm so layer over time after surface application of FGD-CaSO₃ or FGD ourseum

Table 3. Concentrations of selected water-soluble plant nutrients and Al in the surface 10 cm soil layer on day 14 after surface application of FGD-CaSO, or EGD-onesum

t			Treatment		
	Control	FGD-	FGD-	FGD-gypsum	LSD <sub>0.05</sub>
		CaSO <sub>3</sub>	CaSO <sub>3</sub>		
		1.12 Mg	3.36 Mg	1.12 Mg ha <sup>-1</sup>	
		ha <sup>-1</sup>	ha <sup>-1</sup>		
			mg kg <sup>-1</sup>		
	0.202 a	0.016 c	0.097 b	0.067 bc	0.079
	5.8 c	11.3 bc	25.0 a	16.1 b	6.2
	7.7 d	78.7 b	202 a	49.6 c	28.2
	3.1 d	20.4 b	40.7 a	14.5 c	5.3
	3.7 c	85.2 b	209 a	58.7 b	31.0
	0.02 c	0.05 b	0.14 a	0.01 c	0.02
	< 0.001	< 0.001	< 0.001	< 0.001	
	2.547 a	0.170 bc	0.003 c	0.382 b	0.262
	0.067 b	0.088 b	0.431 a	0.005 b	0.122
	0.011 a	0.004 b	0.006 ab	0.001 b	0.006
	0.048 a	0.022 b	0.018 b	0.021 b	0.010
	2.74 a	0.05 b	0.03 b	0.19 b	0.19
				- A. 100	

Com emergence rates were not affected by FGD-CaSO<sub>3</sub> or FGD-gypsum treatment compared with the untreated control at all days measured (Table 4). Sown immediately after treatments (0 day), the corn emergence rate was decreased by the 1.12 Mg ha<sup>-1</sup> FGD-CaSO<sub>3</sub> treatment compared with the 1.12 Mg ha<sup>-1</sup> FGD-gypsum treatment. However, dry weight of corn plants was not affected by FGD-CaSO<sub>3</sub> or FGD-gypsum treatment, compared with the untreated control, when corn seeds were sown on days 0, 1, 2 and 8 after treatments, but it was decreased on days 3 and 14. Why there was not a consistent growth response for all days is not clear but may be due to decreasing availability of plant nutrients such as Mo in the soil (Table 3). Concentrations in corn plant tissue of S were significantly increased by FGD-CaSO<sub>3</sub> or FGD-gypsum. However, concentrations of K, Cu and Mo were decreased. Concentrations of B were increased by 3.36 Mg ha<sup>-1</sup> FGD-CaSO<sub>3</sub> compared with 1.12 Mg ha<sup>-1</sup> FGDgypsum. This indicated that B in the FGD-CaSO<sub>3</sub> was easily taken up by corn plants.

Al from 21-day-old seedlings planted on day 14 after surface application of

CaSO<sub>2</sub>

1.12 Ma 3.36 Ma

47700 ab 47300 ab 44800 b 3270

3070

2670 2660 2790

2960 a

2.69 a 2.40 ab 1.64 b

0.88 ab 0.67 b

31.8 31.8

73.7 68.7 52.1

Means in a row followed by different letters are significantly different at P<0.05.

Treatmen

FGD-

CaSO<sub>2</sub>

3090 a 3150 a

156 154

54.5 55.4

FGD- LSDans

gypsum

1.12 Mg ha

3160

4 30 h

0.55 b 0.40

31.3

lomont

Control EGD-

48200

2830

2580

2070 h

4.75 ab 4.76 ab 5.11 a

283.a

1.08 :

25.9

105.6

Treatme	ent	Days after treatment								
Material	Rate Mg ha <sup>-1</sup>	0	1	2	3	8	14			
Emergence rate (%)										
Control	0	86.5 ab	84.6	94.2	92.3	86.5	94.2			
FGD-CaSO <sub>3</sub>	1.12	73.1 b	96.2	96.2	84.6	88.5	92.3			
	3.36	86.5 ab	94.2	98.1	88.5	90.4	90.4			
FGD-gypsum	1.12	100 a	90.4	92.3	88.5	84.6	90.4			
LSD <sub>0.05</sub>		26.4	12.5	11.4	12.6	16.0	12.3			
				Growth	(g/plot)					
Control	0	7.35	6.75	9.66	10.37 a	10.03	9.22			
FGD-CaSO <sub>3</sub>	1.12	5.45	7.71	9.31	8.18 b	8.68	6.50			
	3.36	5.90	7.33	9.84	9.35 ab	7.73	7.51 a			
FGD-gypsum	1.12	6.33	6.84	9.12	8.39 b	8.24	7.72 a			
LSD <sub>0.05</sub>		2.57	1.98	4.00	1.85	2.80	2.36			

#### CONCLUSIONS

Suffite in FGD-CaSO<sub>3</sub> was rapidly oxidized to SO<sub>4</sub><sup>2</sup> when applied to the moist surface of a field in the spring. Effects of FGD-CaSO<sub>3</sub> on water-soluble K, Al, Fe, Mn, and Zn in soils, on corn emergence and growth, and on concentrations of elements in plant tissue were similar to those of FGD-gypsum when applied at the rate of 1.12 Mg ha<sup>-1</sup>, FGD-CaSO<sub>3</sub> provided more Ca. Mg and B to the soil than FGDgypsum did. It is safe for plants if FGD-CaSO<sub>3</sub> is applied to the field surface.

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1340

583

294

720

0.62

0.91

13.1

10.5

71.3

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