

A Rapid and Cost Effective Soil Carbon Mineralization Method For Static Incubations L.A Sherrod, J.D. Reeder, Jim Hunter, and L.R. Ahuja USDA-ARS, Fort Collins, Colorado



INTRODUCTION

Soils can be a net carbon sink if C input is greater than C output (Organic C mineralized to CO₂).

Soil C mineralization by the biological oxidation of organic C in soil is one of the main neans C is returned to the atmosphere.

Management and climate have a direct and relatively quick impact on biologically active C pools.

Soil incubations with subsequent determination of CO2 are becoming common oil assays used to determine active soil organic C (SOC) pools.

C mineralization rates and soil microbial biomass are also used to evaluate soil quality.

 Traditional methods to determine C nineralized in a static incubation are costly in ooth labor and equipment.

A rapid and cost effective method for determining C mineralization rates would allow nore routine evaluation of this dynamic pool in both research and in routine soil testing labs.

OBJECTIVE

)biectives: Evaluate if the single cell infrared gas analyzer (IRGA), gas chromatograpy (GC) and NaOH basetrap titration methods of CO2 determination obtain the same concentrations under static soil incubations over 1, 3, 10 and 21 days.

Evaluate how well these methods correlate to each other across incubation times.

Estimate the limit of quantitation (LOO). Estimate the concentration of CO.

and O₂ gases in the incubation chamber headspace that suppres respiration.

METHODS

ass a 2-mm sieve.

luplications for the 3 detection

nethods of IRGA, GC, and alkali

NaOH) and brought to 50% water

bsorption followed by titration

illed pore space (n=96)

Surface soil samples were collected rom a catena sequence from 4 ropping systems across 3 slopes and 2 ield replications to obtain a range of abile organic matter (n=24).

ented soils were returned to ambient CO2 and O2 levels Soils were air dried and ground to between readings by inserting hypodermic needles and sparging with compressed air. Soils were weighted out to 30-g with 4

Part 2:

ize of GC and NaOH.

Part 1:

septa port.

Oxygen was monitored concurrently with CO₂ by linking the output port of the IRGA analyzer to the input port of O, analyzer.

Incubations that had a 30-g sample size were continued up o 50 days with CO2 and O2 levels analyzed. On day 35 a 100-mL injection of 100% O2 was injected to see if CO2 oncentration increased.

IRGA soils were incubated in 500-mL Wheaton serum ottles sealed with a rubber septum and aluminum ring.

GC and Alkali absorption (NaOH) soils were incubated

in 1-L mason canning jars with soils weighted into 60-mL

nap jars. Alkali absorption also had a 20 mL vial of 1N

NaOH within the canning jar. The lids of the canning

ars used for GC analysis were modified by inserting a

Bulk density value of 1.0 was artificially obtained by

apping soils into a known volume within incubation

oottles. Duplicates were averaged before statistical

Base traps were switched out on 1, 3, 10 and 21 day

IRGA and GC headspace gases were sampled using a

pressure-lok gas syringe sequentially over 1, 3, 10, and 21

lay period. The IRGA bottles were vented after the 10

day measurement to mitigate the potential for respiration

uppression as the incubation chamber volume was 1/2 the

Determination of headspace CO, and O, gas levels that

show suppression of respiration was done over a 30 day

period for each 15-g and 30-g sample size in 500-mL

Wheaton serum bottles incubated at 30°C with the

verage of 24 soils calculated over each time period.

ncrements and titrated to a nH endpoint of 8.3.

nalysis to normalize the sub-sampling variability (n=24).

RESULTS – PART 1



Soils incubated in the 500-mL bottles showed a lower level of respired C at day 10 than the 1-L mason jars. After the 10 day reading IRGA soils were vented and did regain values similar to GC and NaOH at day 21.

Comparison of CO. IRGA and NaOH Comparison of CO, from GC vs. NaOH

v - 1.0550v

, R² = 0.8035





All methods were strongly correlated with r² values of 0.80 and above. The strongest slope 1.05) was found between the regression of NaOH vs. GC, both of which had the larger 1-L cubation chamber

RESULTS – PART 2



·Limit of Ouantitation is defined as the lowest level at which a analytical asurement becomes meaningful in quantifying a result. The empirical value for LOQ is found by examining the inflection point in the above curves of relative standard deviation vs. increasing analyte concentration. The lowest LOQ was found with the GC method of 90 ul CO./L followed by 115 for IRGA. The NaOH ethod had the highest LOO of 600 ul CO₂/L.



Mean of 24 soils over 3, 10, 21, and 30 days of 15-g vs. 30-g sample size confirmed that respiration was being suppressed at about day 10 in the 500-mL serum bottle measured with the IRGA

•Day 10 is, of course, sample dependent, therefore the headspace gas oncentration that shows suppression of respiration is required to estimate when the incubation bottles need to be vented back to ambient levels of CO, and O.





•The graph above shows the average of 24 soils % CO, and O, in he headspace above a 30-g sub-sample within a 500-mL serum bottle sealed with septa and Al sealing ring. The level of CO₂ and O₂ that showed a slope change and thus suppression where found o be 9.1% and 10.9 % respectively.

At day 35, 100 mL of 100% O, gas was injected into the incubation bottles. This restored ambient O2 levels but not CO2 levels. This would indicate that the suppression was due to limited O₂ levels not toxic CO₂ levels as CO₃ levels in the headspace started climbing after day 35.

CONCLUSIONS

•Long-term incubation are achieved by venting the 500 mL serum bottles between readings and/or monitoring the headspace gases to keep CO, levels below 10% and O, levels above 10%.

•Reduction of the incubation vessel volume by half (IRGA only) showed that venting the headspace is needed in longer incubations. Therefore results at 21 days showed a single-cell IRGA is sufficient technology to assay respired CO2 from static incubations as results are not statistically different than the gold standard of GC analysis and the traditional alkali absorption followed by titration.

•The sample run time for the IRGA method using out of the box software is 90 or more per hour vs. 26 for the GC and under 10 per hour for base-trap titration (NaOH).

•Equipment cost favor the IRGA method. A single cell IRGA is approximately 1/10th the cost of a GC or automated titrator.

•Equipment costs, simplicity and run times make the single cell IRGA method suitable for research and routine soil anal