# **Governing Controls of Sulfur On Arsenic Uptake By Rice in Paddy Soil**



#### Background

Arsenic (As) uptake in rice from contaminated soil and irrigation water is currently being highlighted as an important exposure pathway for humans to this potent toxin<sup>1</sup>. Sulfur (S) has a high affinity for arsenic, both in minerals and organic molecules, and the addition of excess S can decrease the uptake<sup>2,3</sup>, concentration<sup>2,4</sup> and translocation<sup>2,4</sup> of As in rice. However, the exact mechanisms behind these effects are not fully understood.

### Aim and approach

We combine monitoring and observation of bulk effects with in-depth analyses of microscale processes in the rice rhizosphere, intending to elucidate the governing controls of sulfur on arsenic mobility, uptake and translocation in the paddy rice system with and without organic amendments, as conceptualized in Fig. 1.

Table 1. Soil properties							
Texture	Soil K Sandy Clay Loam	Soil P		Table 2. Amendment total element concentration			
рН	6.13	5.39		concentration			
Totals					Dry	Charred	Dry
As (µg/g)	15	11			Husks	Husks	Strav
Fe (mg/g)	45	44		C (%)	37	49	37
S (μg/g)	330	363		N (%)	0.2	0.4	0.9
C (%)	1.3	1.6	/	S (µg/g)	26	21	192
N (%)	0.14	0.17		As (µg/g)	0.02	0.08	0.06
Oxalate extractable							
As (µg/g)	11	6.6					
Fe (mg/g)	21	24					

#### Materials & Methods

Material Two Cambodian paddy soils (Table 1) Four organic amendments plus gypsum (Table 2) Experiments Pot trials with/without rice plants Batch reactors Analytical methodology Solids – XRF, XAS, NMR Solution - ICP, IC, HPLC-ICP-MS, TOC, Spectrophotometry Rhizosphere - µXAS, SEM, TXM, Microsensors Microbiology - Isothermal microcalorimetry, Pyrosequencing

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References <sup>1</sup> Halder et al. (2012) Environ. Sci. Technol., 46, 4142–4148 <sup>2</sup> Fan et al. (2013) Plant Soil Environ., 59, 169-174

# **Results – solubility coupled to microbial reductive dissolution**

Dried rice straw and charred rice straw increased the rate of As release from soil K in a batch experiment (Fig. 2). The effect was most prominent with the dried straw, due to an almost 5.5 times higher C addition.

However...

- The charred straw treatment had not reached equilibrium by the end of the batch experiment (after 65 days).
- The microbial activity (Fig. 3) was lower with charred straw, even when the application rate was adjusted to add the same amount of C, suggesting that the differences can be attributed to C chemistry rather than amount.



Figure 2. Arsenic and iron dissolution/ desorption rates (% of total element in soil, i.e. 15 µg As/g soil, 45 mg Fe/g soil) in batch reactors with 4 g of soil K, flooded with 125 ml TRIS buffer (10 mM, pH 7.0), amended with 4% dried or 2% charred rice straw, gypsum or unamended. The amendment additions were set to 7.7 µg S added per g soil. All vials were sealed immediately after flooding and placed on a shaker in a dark incubator (25 C). Each sampling withdrew a proportional (soil:solution) sample of 15 ml.



# **Results – solid phase speciation**

Flooding caused arsenic reduction in both soils and increased the relative contribution from As(III) species from 30% to 51% and 19% to 53% in soil P and K respectively (Fig. 4a). Although the flooding also increased the contribution from sulfides to the total sulfur speciation (Fig. 4b), the fitting of the As XANES spectra did not indicate any formation of As-S species after 53 days of flooding.





Figure 3. Microbial activity, estimated by a) heat output (J/g soil) (using isothermal micro-calorimetry) and b)  $O_2$ depletion/CO<sub>2</sub> production (using in situ optical spot sensors), in 4 g of dry soil K, flooded with 15 ml TRIS buffer (10 mM, pH 7.0), amended with dried or charred rice straw or un-amended. All vials were sealed immediately after flooding. Two application rates were used for the organic amendments in the 150 🗟 microcalorimetry measurements (Fig. 3a): 1 – same amount of S (7.7 µg S/g soil, as in batch experiment), 2 same amount of C (9 mg C/g

ate	Rel. %	Red. Inorg. S	Red. Org. S	Inter- med. S	Ox. S			
	P2 —	10	66	8	17			
·	P0 —	7	42	9	42			
	к2 —	14	40	3	43			
	ко —	8	43	7	42			
2485 2490	speci after in un	ation (2) 53	in soi 8 days	ls P a of flo	nd I odii	< bef ng wi	sulfur ore (0) ith DI w th 1% d	and ater