

Efficiency of the nitrification inhibitor 3,4-dimethylpyrazole phosphate (DMPP) to mitigate N₂O emissions as affected by environmental variables and soil properties

INTRODUCTION

- Synthetic N fertilizers and manure applied to soil are the main sources of nitrous oxide (N₂O) emissions in agriculture.
- Nitrification inhibitors (NIs) reduce N₂O emissions from manure by suppressing the activity of ammonia oxidising bacteria and possibly archaea, thereby delaying the conversion of NH₄⁺ to NO₃⁻.
- Efficiency of NIs depend on environmental and edaphic factors such as soil temperature, moisture content, pH, texture and soil NO₃⁻.

OBJECTIVES AND HYPOTHESES

To study how soil texture, temperature, NO₃⁻ availability and manure-soil contact affect N₂O mitigation by DMPP.

H₁: Potential of DMPP to reduce N₂O emissions would be higher at 10 compared to 20 °C due to slower degradation of DMPP

H₂: Shallow mixing of manure (MM) with soil will lead to higher N₂O emissions than surface application (MS) by enhancing nitrification and hence efficiency of DMPP will be higher in the MM treatments.

H₃: N₂O emissions from soil with low pH (CS_{4,8}) will be higher than emissions from neutral-pH soil (CS_{6,4}); the effect of DMPP to mitigate N₂O emissions will depend on soil NO₃⁻ availability.

METHODOLOGY

- Three different incubation experiments were carried out to investigate effects and interactions of selected soil properties with respect to N₂O emissions from cattle manure and mitigation by DMPP.
- Soil was packed into 100 cm³ stainless steel rings in four portions, each time with addition of water to each layer (in some treatments with NO₃⁻) to the predetermined WFPS.
- Manure ±DMPP was either surface-applied to the central 50% of the ring surface or (as part of Experiment 2) mixed into the top 1 cm of the repacked soil sample.
- After manure application, N₂O fluxes were determined at regular intervals after manure application for a period of one month.
- Temporal and spatial dynamics of mineral N, pH, organic matter content etc. in three depth intervals were determined by the end (Experiment 1) or 4-5 times during the experiment by destructive sampling.
- In Exp. 3, on days 7, 14 and 28, soil samples from 1-5 mm depth were analysed for nitrifier and denitrifier gene abundances using real-time quantitative PCR.

	Experiment 1		Experiment 2		Experiment 3	
Experimental conditions						
Soil texture	Sandy loam	Coarse sandy	Sandy loam	Sandy loam	Coarse sandy	Coarse sandy
WFPS (%)	55	55	55	55	25	25
Temperature (°C)	10/20	10/20	20	20	20	20
Soil pH	6	6.4	6.4	6.4	6.4	4.8
Applications						
Manure application	Surface	Surface	Mixed*	Surface	Surface	Surface
Nitrate application (mg g ⁻¹)	0	0	0	0	0/50	0/50

Table 1: Factorial experimental design and parameters

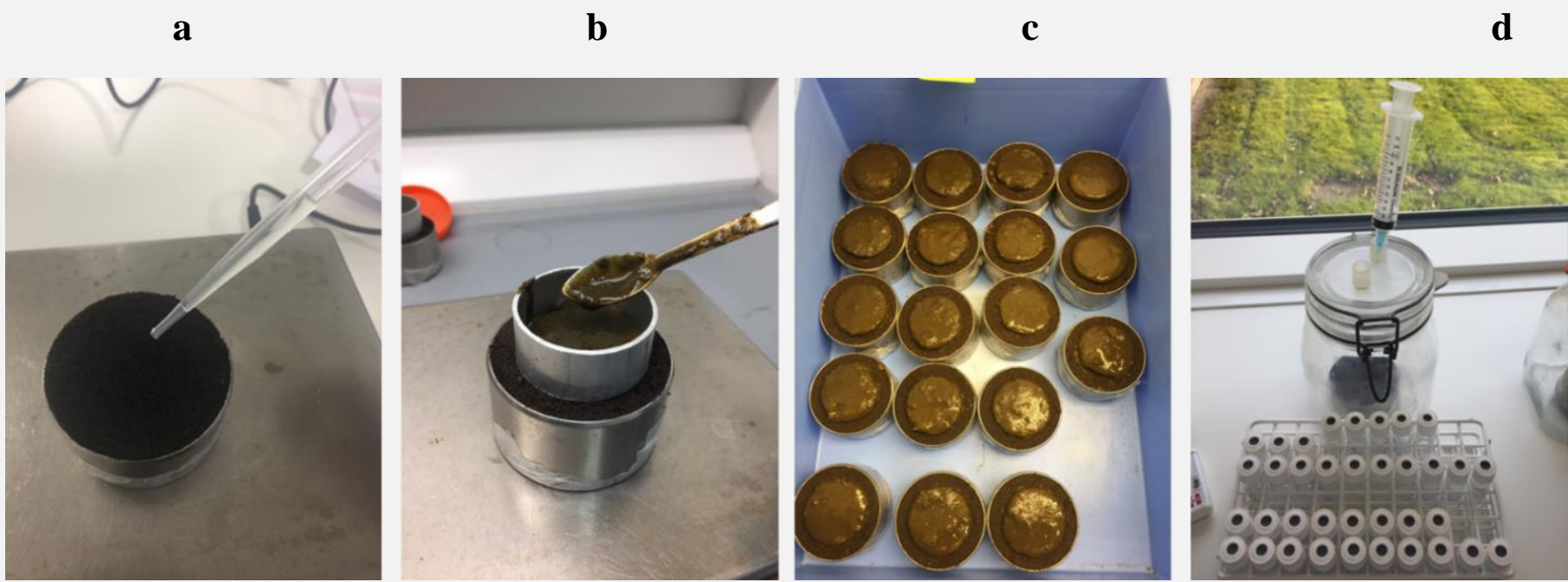


Fig 1: Water and manure addition to packed rings (a,b,c). Rings placed in 1L jar for N₂O sampling



Fig 2: Soil slicing procedure: a) customized instrument for holding and slicing of soil columns b) slicing of top layer c) sliced layer d) top layer with manure separated for analysis of pH and moisture

RESULTS

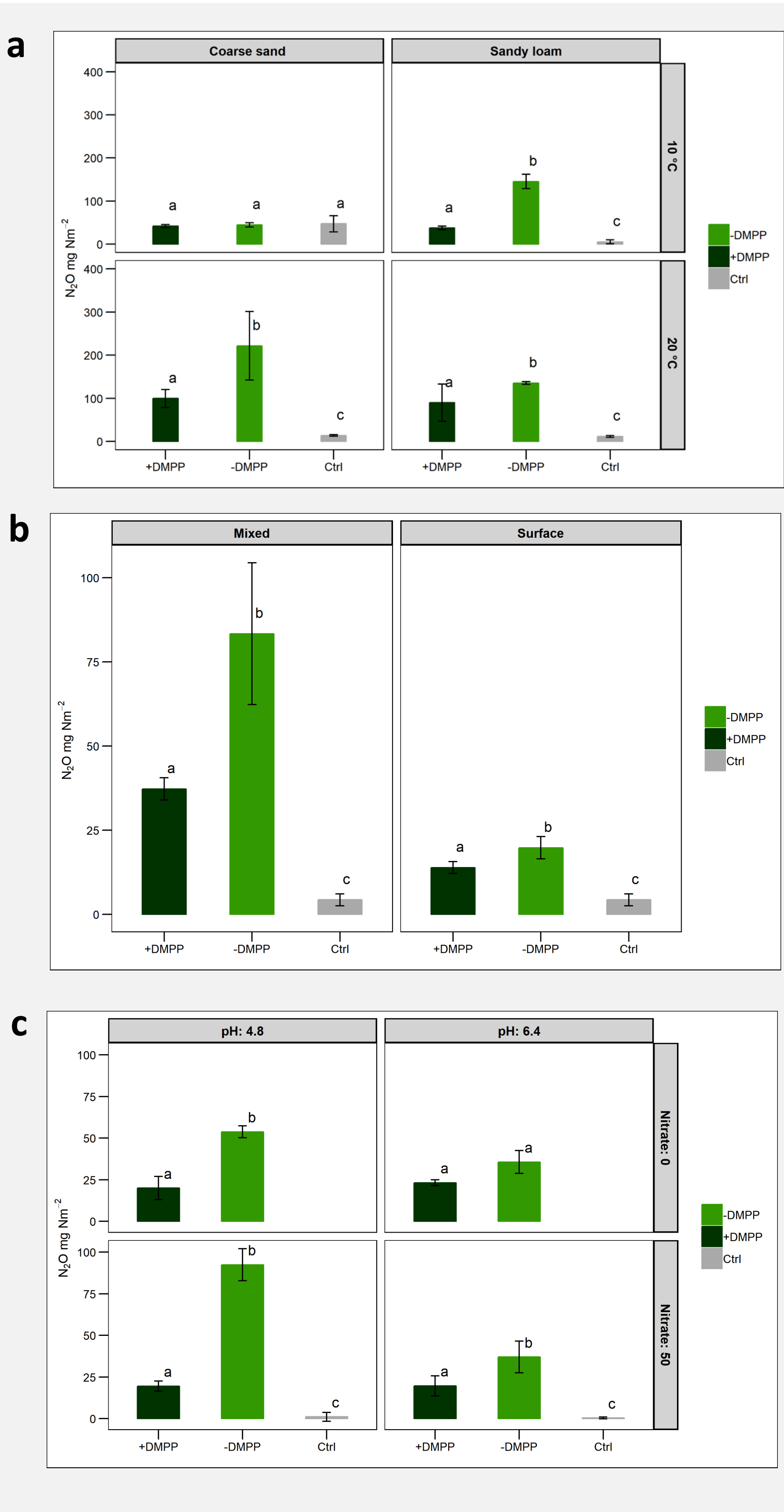


Fig 3: Cumulative N₂O emissions from Experiment 1 (a), 2 (b) and 3 (c)

RESULTS (CONTINUED)

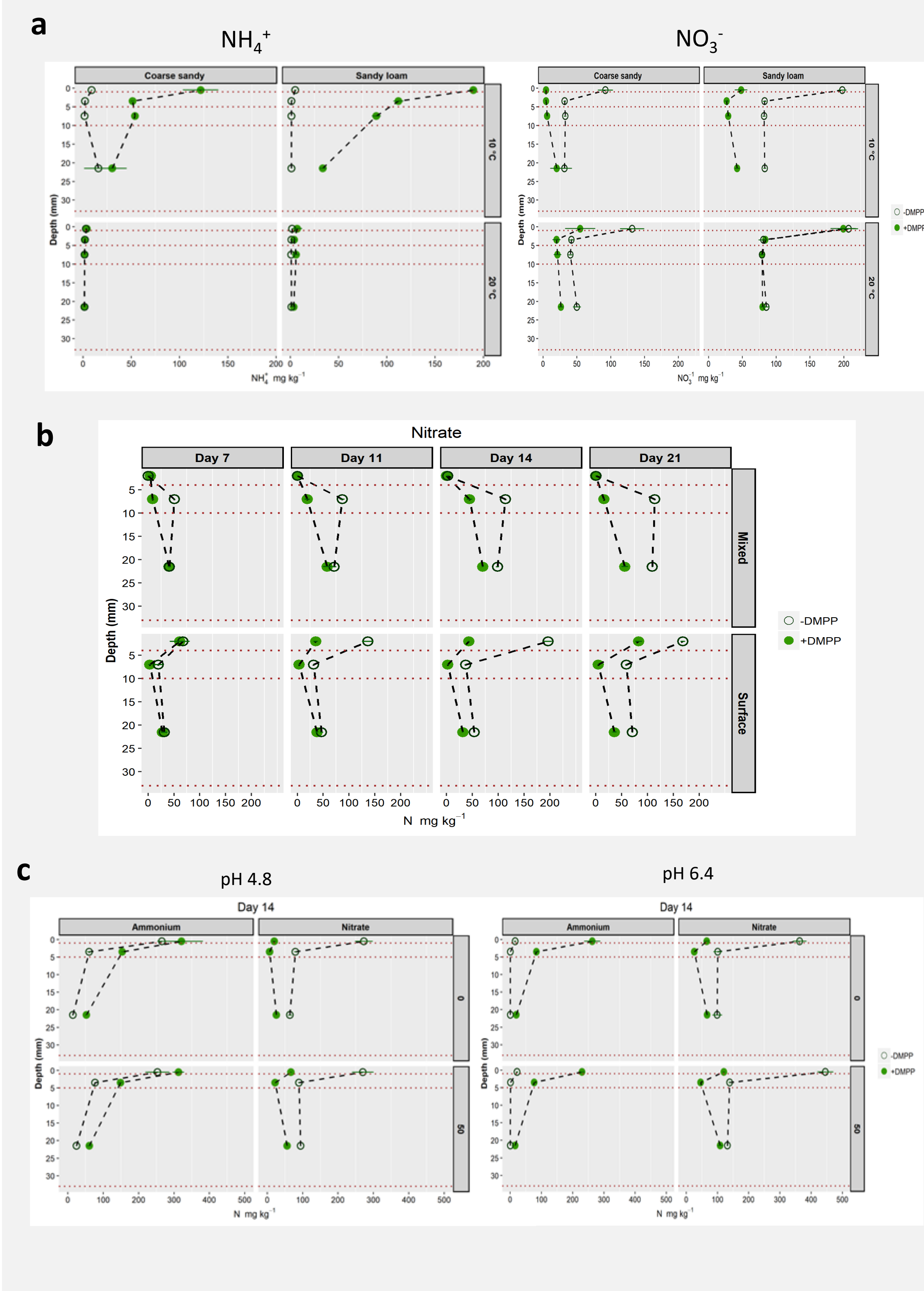


Fig 4: Temporal and spatial mineral N dynamics from Experiment 1 (a), 2 (b) and 3 (c)

RESULTS (CONTINUED)

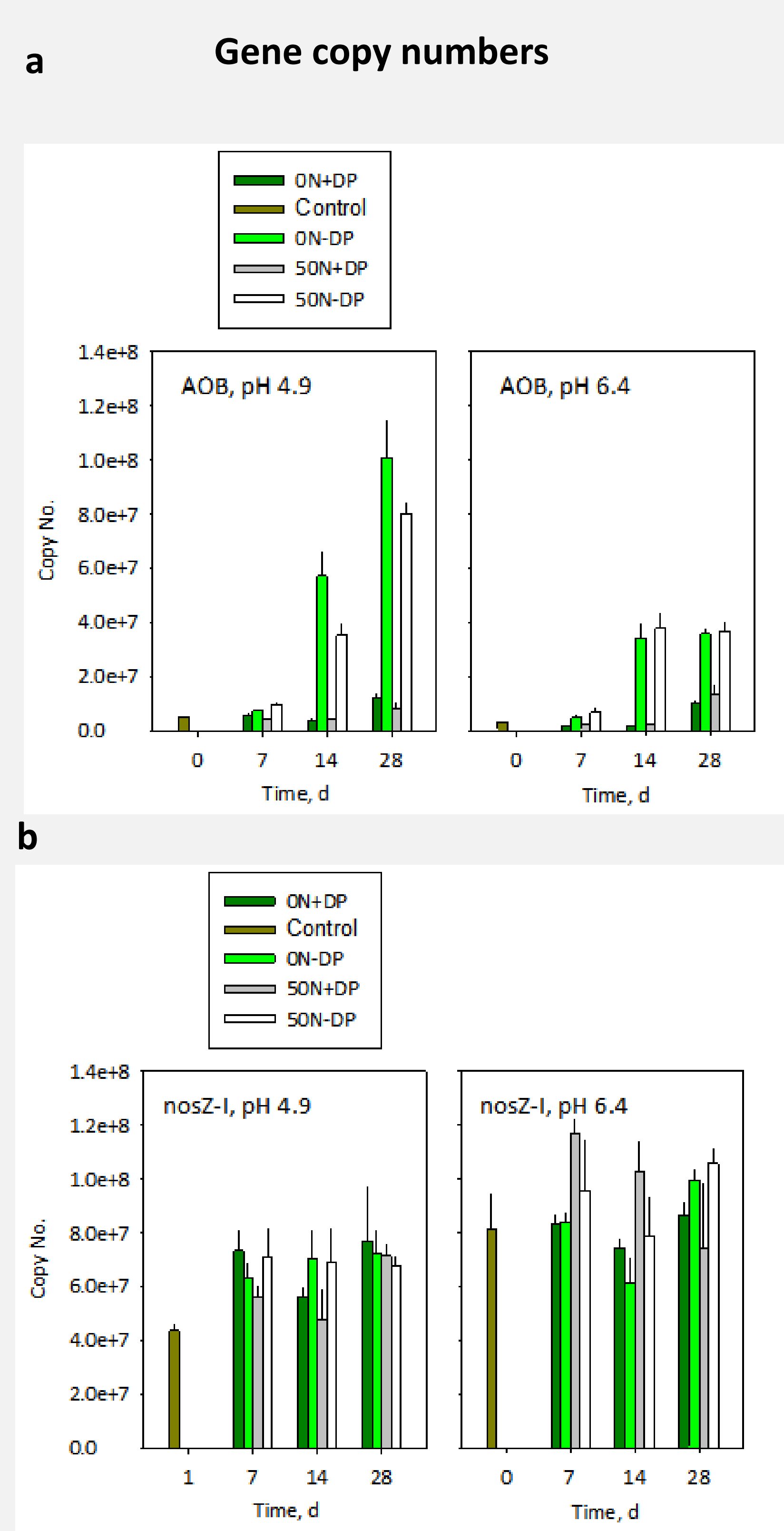


Fig 5: Nitrifier and denitrifier microbial gene abundance as affected by soil pH, NO₃⁻ and DMPP

CONCLUSIONS

- Exp 1: Efficiency of DMPP was influenced by both temperature and soil texture. In sandy loam soil, DMPP reduced N₂O emissions more at 10°C than at 20°C, while no reductions were observed at 10°C in coarse sandy soil (Fig 3a)
- Exp 1: NO₃⁻ accumulation was lower in coarse sandy soil at 10 °C indicating delayed nitrification (Fig 4a)
- Exp 2: In general, N₂O emissions were higher in treatments with manure mixed to 1-10 mm of soil and efficiency of DMPP was independent of application depth.
- Exp 2: The depletion of soil NO₃⁻ at 0-5 mm when manure was mixed into the top soil, but accumulation of NO₃⁻ when surface applied, indicates that better soil-manure contact enhanced denitrification. At 5-10mm, opposite trends were observed
- Soil pH was an important factor that controlled N₂O emissions and higher emissions were observed in acidic soil (CS_{4,8}) attributed to lower abundance of nosZ genes resulting in higher N₂O:N₂+N₂O (Fig 5b).
- DMPP reduced N₂O emissions in all treatments irrespective of nitrate availability in CS_{4,8} and only in treatments with NO₃⁻ amendment in CS_{6,4} (Fig 3c). In contrast to the hypothesis, N₂O mitigation potential of DMPP was found to be higher when NO₃⁻ availability was not a limiting factor across both soil pH (Fig 3c).
- In comparison to CS_{4,8}, NO₃⁻ accumulation was higher in CS_{6,4} on day 14 indicating increased nitrification rates (Fig 4c)
- DMPP reduced AOB abundance significantly in all treatments (Fig 5a), but had no effect on the abundance of AOA (data not shown).
- In CS_{4,8} DMPP reduced the abundance of ComaA (clade A), ComaA ntsp, as well as nosZ-I

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